

Corrigendum

Corrigendum to “A trace metal clean reagent to remove surface-bound iron from marine phytoplankton”
[Marine Chemistry 82 (2003) 91–99][☆]

Antonio Tovar-Sanchez^{a,*}, Sergio A. Sañudo-Wilhelmy^a, Manuel Garcia-Vargas^b,
Richard S. Weaver^c, Linda C. Popels^c, David A. Hutchins^c

^a Marine Sciences Research Center, Stony Brook University, Stony Brook, NY 11794-5000, USA

^b Department of Analytical Chemistry, University of Cadiz, Poligono Rio San Pedro s/n, Puerto Real, Cadiz, Spain

^c College of Marine Studies, University of Delaware, 700 Pilottown Rd., Lewes, DE 19958, USA

The authors regret that in the above article an error occurred in Table 1. The correct version is printed below.

Table 1
Composition and preparation of the oxalate reagent

Chemicals ^a	Basic oxalate reagent	Trace metal clean oxalate reagent	Concentration (mol/L)
(a) EDTA (EDTA-Na ₂ ·H ₂ O)	1.86 g	2.23 g	0.05
(b) Sodium citrate (C ₆ H ₅ Na ₃ O ₇ ·2H ₂ O)	1.47 g	1.76g	0.05
(c) KCl	0.074 g	0.089 g	0.01
(d) NaCl	0.5 g	0.6 g	0.1
(e) NaOH (10 M)	drops until pH 6-7		
(f) Oxalic acid (C ₂ H ₂ O ₄ ·2H ₂ O)	1.26 g	1.51 g	0.1
(g) NaOH (10 M)	buffer to pH 8		
(h) Hydroxylamine (1.44 M, NH ₂ OH·HCl) ^b	–	0.5 ml	7.2e – 3
(i) Perchlorate (0.008 M, NaClO ₄ ·H ₂ O)	–	6.5 ml	5.2e – 4
(j) 1,10 phenanthroline (0.055 M, C ₁₂ H ₈ N ₂ ·H ₂ O) ^c	–	13 ml	7.2e – 3

^a To prepare the basic, non-trace metal clean reagent, mix consecutively from (a) to (f) in 60 ml of MQ water and take to 100 ml with MQ water after (g); for the trace metal clean oxalate reagent continue to (j) and follow the procedure outlined in the text to a final volume of 120 ml.

^b Hydroxylamine solution is buffered to pH8 with NaOH.

^c Several drops of sulfuric acid are added to completely dissolve solution.

[☆] doi of original article S0304-4203(03)00054-9.

* Corresponding author. Tel.: +1-631-632-6913; fax: +1-631-632-8820.

E-mail address: atsanchez@notes.cc.sunysb.edu
(A. Tovar-Sanchez).